# Universida<sub>de</sub>Vigo

Subject Guide 2023 / 2024

IDENTIFYIN				
Geology: G				
Subject	Geology: Geology			
Code	V11G201V01106			
Study	Grado en Química			
programme				
Descriptors		Choose	Year	Quadmester
	6	Basic education	1st	2nd
Teaching	#EnglishFriendly			
language	Spanish			
	Galician			
Department				
	Gago Duport, Luís Carlos			
Lecturers	Gago Duport, Luís Carlos			
	Gil Lozano, Carolina			
E-mail	duport@uvigo.es			
Web	http://moovi.uvigo.gal/login/index.php			
General description	The study of the structure of matter in a crystalline state the understanding of the most diverse phenomena in the the Earth as a geochemical system, the Approach of the degree in Chemistry is mainly oriented towards the student mechanisms. These topics are approached from the poin Geochemistry. Starting from the thermodynamic and king crystalline phases, structural aspects, crystallographic mimportance of these processes is introduced for the student such as semiconductors, pharmaceuticals, biological materials and bibliographic references in English, b) to in English.	e field of Chemist subject Geology by of crystalline stint of view of Crystalline stinetic mechanisms notation and diffrady of natural (mine acromolecules, and uest from the teac	ry, therefore, after corresponding to to ructures and cryst allography, Minera that lead to the foction are studied. eral) crystals and so d ceramic materia	r a general view of the first year of the callization alogy and ormation of As a corollary, the synthetic materials, Is, among others.

## Training and Learning Results

Code

- A2 Students have demonstrated knowledge and understanding in a field of study that builds upon their general secondary education, and is typically at a level that, whilst supported by advanced textbooks, includes some aspects that will be informed by knowledge of the forefront of their field of study
- A3 Students have the ability to gather and interpret relevant data (usually within their field of study) to inform judgments that include reflection on relevant social, scientific or ethical issues
- B1 Ability for auronomous learning
- B3 Ability to manage information
- B4 Ability for analysis and synthesis
- C4 Use computer tools properly to obtain information, process data, perform computational calculations and calculate matter properties
- C9 Know the structural aspects of chemical elements and their compounds, including stereochemistry
- C10 Know the characteristics of the different states of matter and the theories used to describe them
- C15 Know the main techniques of structural research, including spectroscopy
- C16 Know the relationship between macroscopic properties and properties of individual atoms and molecules, including macromolecules (natural and synthetic), polymers, colloids, crystals and other materials
- D3 Ability to communicate in both oral and written form in Spanish and / or Galician and / or English

Expected results from this subject					
Expected results from this subject	Training and Learning				
	Results				
Describe and explain the Earth as a system.	A2 B4 C10 D3				
	A3 C16				

		C10 C15 C16	
		C16	
		0-0	
		C9	
		C10	
		C16	
A2		C9	
		C10	
		C15	
		C16	
A3	В3	C9	D3
	B4	C15	
		C4	
		C9	
		C10	
		C16	
A2	B1	C4	
	В3	C15	
	В4		
_		A2 B1 B3	B4 C15 C4 C9 C10 C16 A2 B1 C4 B3 C15

Contents	
Topic	
The Earth as a Geochemical System: Rocks	Geochemical evolution of the Earth. Plate Tectonic. The rocks cycle.
forming minerals	Comparison of Earth with other planets in the solar system: The case of of
-	Mars.
The crystallization process: thermodynamic and	Differences between nucleation and crystal growth.Crystal growth kinetics.
kinetic aspects.	Structural aspects.
Characterization of crystalline solids: structure vs	s.Microscopic and macroscopic approaches to crystalline solids
morphology.	
Isotopes in Geology: Measuring the geological	Radioactive isotopes and stable isotopes. Isotopic dating techniques. The
time with radiactive isotopes.	Isochrone method. Kinetic tracking of processes using stable isotopic
Analyzing kinetic processes by fractionation of	techniques. Notation and units. Rayleigh fractionation.
stable isotopes.	,
Geometric crystallography: Periodicity and	Two-dimensional lattices. Point symmetry. Schoenflies and Hermann-
symmetry in the crystals.	Mauguin notations of point symmetry elements and clases. Bravais
	lattices. Microscopic symmetry Space groups. Miller indices and zone axes.
	Fractional coordinates
X-ray crystallography: Bragg's Law and the Phase	e The physical basis of diffraction. Diffraction by crystals lattices and
problem	radiation sources. The Bragg Law The reciprocal lattice. Diffraction
	Patterns. Indexing of diffraction diagrams, powder diagrams and
	monocrystal diagrams Quantitative Analysis. The Phase problem. Methods
	of resolution of structures from diffraction datasets.

Planning				
	Class hours	Hours outside the classroom	Total hours	
Lecturing	26	70	96	
Mentored work	1	5	6	
Laboratory practical	6	0	6	
Problem solving	6	34	40	
Objective questions exam	1	0	1	
Problem and/or exercise solving	1	0	1	

<sup>\*</sup>The information in the planning table is for guidance only and does not take into account the heterogeneity of the students.

Methodologies	
	Description
Lecturing	First, the basic principles of crystallization are analyzed from a geological and thermodynamic point of view. Further, we introduce to the student the fundamentals of isotope geochemistry.  Next themes are devoted to the structural characterization of crystals, analyzing the concepts of periodicity and symmetry in 2D and 3D crystalline lattices.  Finally we introduce the principles and practical aspects of diffraction techniques applied to the estructural analysis
Mentored work	A short paper will be write by groups summarizing the laboratory work. Some guidelines concerning formats and content will be given before the realization.  A seminar will be assigned to guide each working group in this task.

Laboratory practical	They will be dedicated to the study of the crystallization process, analyzing three aspects:
	(1) Crystallization in nature: mineral recognition
	(2) Optical microscopy with polarized light.
	(3) Crystal growth from solutions and silica gels in the laboratory.
Problem solving	Seminars will be used to solve practical exercises about nucleation and crystal process and to solve issues related to crystallographic notation and concepts

Personalized assistance			
Methodologies Description			
Problem solving	The resolution of exercises will be carried out during the seminars by answering to the questions raised in class.		
Mentored work	They will be developed in the computer classroom and in theoretical class as well as through the realization of tutorials or consultations using the Tema platform or the electronic mail.		

Assessment			
	Description	QualificationT	raining and
			Learning
			Results
Mentored work	The completion of a report/work whose content will be related to the activity	10	
	carried out in the laboratory and seminars will be evaluated.		
Laboratory practical	The activity carried out in the mineralogy laboratory and in the completion of	30	
	exercises and questionnaires proposed through the Moovi platform will be		
	evaluated.		
Problem solving	The exam will also include	20	
	exercises and/or problems.		
Objective questions	Exam with short questions and multiple choice questions, as well as a topic	40	
exam	about the content of the theoretical classes and/or seminars.		

#### Other comments on the Evaluation

## Sources of information

### **Basic Bibliography**

Andrew Putnis, Introduction to Mineral Sciences, 6ª, Cambridge University Press, 2008

Edward Tarbuck y FredericK Lutgens, Ciencias de la Tierra. Una introducción a la Geología Física, 10ª, Pearson, 2013

**Complementary Bibliography** 

Christofer Hammond,, The Basic of Crystallography and Diffraction, 3ª, Oxford University Press, 2009

Jose Luis Amorós, La gran aventura del cristal, 1ª, Ediciones Complutense, 2017

Carmelo Giacovazzo et al., **Fundamentals of Crystallography**, 2ª, Oxford University Press,

## Recommendations

## Subjects that continue the syllabus

Chemistry: Chemistry 2/V11G201V01109

### Subjects that are recommended to be taken simultaneously

Physics: Physics 2/V11G201V01107

Mathematics: Mathematics 2/V11G201V01108 Chemistry: Chemistry Lab II/V11G201V01110 Chemistry: Chemistry 2/V11G201V01109

## Subjects that it is recommended to have taken before

Biology: Biology/V11G201V01101 Physics: Physics I/V11G201V01102

Mathematics: Mathematics 1/V11G201V01103 Chemistry: Chemistry Lab I/V11G201V01105