Universida_{de}Vigo

Subject Guide 2015 / 2016

			Su	bject Guide 2015 / 2016
IDENTIFYIN	IG DATA			
Degree the				
Subject	Degree thesis			
Code	V11G200V01991			
Study	(*)Grao en			
programme	Química			
Descriptors	ECTS Credits	Choose	Year	Quadmester
	18	Mandatory	4th	2nd
Teaching				
language				
Department Coordinator				
Lecturers	Pérez Juste, Ignacio Pérez Juste, Ignacio			
E-mail	uviqpipj@uvigo.es			
Web	http://quimica.uvigo.es/decanatoquimica/traballo-f	in-de-grag html		
General	According to the memory of the Degree in Chemist		of Vigo the End	of Degree project is a
description	mandatory subject of 18 credits ECTS in the secon The objective of the subject is to offer the students competences adquired during the Degree studies. The TFG is an original work that each student will of TFG subjects can correspond to experimental and/	d term of the fourth s the opportunity to do individually under or theoretical works	course. apply the knowle the supervision and/or of bibliog	dges, skills and of one or two tutors. raphic reviews on
	subjects related with the contains in the Degree in written report and its public presentation.	Chemistry. The fina	l stage of the TF	G will consist in a
educati	ts have demonstrated knowledge and understanding ion, and is typically at a level that, whilst supported l ed by knowledge of the forefront of their field of stud	by advanced textboo		
A2 Student or voca	ts can apply their knowledge and understanding in a tion, and have competences typically demonstrated ns within their field of study	manner that indicat		
A3 Studen	ts have the ability to gather and interpret relevant d clude reflection on relevant social, scientific or ethica		neir field of study) to inform judgments
	ts can communicate information, ideas, problems an		pecialist and nor	n-specialist audiences
A5 Studen high de	ts have developed those learning skills that are nece gree of autonomy	essary for them to co	ntinue to undert	ake further study with a
chemic	strate knowledge and understanding of essential fac al terminology, nomenclature, units and unit conver	sions.		· ·
reaction	strate knowledge and understanding of essential fac ns and its main characteristics			
quantu	strate knowledge and understanding of essential fac m mechanics and its application in the description of strate knowledge and understanding of essential fac	f the structure and p	roperties of aton	ns and molecules
solving	strate knowledge and understanding of essential fac analytical problems and characterization of chemica strate knowledge and understanding of essential fac	al substances		
differer	strate knowledge and understanding of essential fac at states of matter and the theories used to describe	them		
thermo	strate knowledge and understanding of essential fac dynamics and their applications in chemistry			• •
includir	strate knowledge and understanding of essential fac ng catalysis and reaction mechanisms			-
	strate knowledge and understanding of essential fac ral determination, including spectroscopy	ts, concepts, princip	ies and theories:	main techniques for

- structural determination, including spectroscopy Demonstrate knowledge and understanding of essential facts, concepts, principles and theories: characteristic properties of the elements and their compounds, including group relationships and variations in the periodic table C9

- C10 Demonstrate knowledge and understanding of essential facts, concepts, principles and theories: properties of aliphatic, aromatic, heterocyclic and organometallic compounds
- C11 Demonstrate knowledge and understanding of essential facts, concepts, principles and theories: nature and behavior of functional groups in organic molecules
- C12 Demonstrate knowledge and understanding of essential facts, concepts, principles and theories: structural features of chemical elements and their compounds, including stereochemistry
- C13 Demonstrate knowledge and understanding of essential facts, concepts, principles and theories: main synthetic routes in organic chemistry, including interconversions of functional groups and the formation of carbon-carbon and carbonheteroatom bonds
- C14 Demonstrate knowledge and understanding of essential facts, concepts, principles and theories: relationship between macroscopic properties and properties of individual atoms and molecules, including macromolecules
- C15 Demonstrate knowledge and understanding of essential facts, concepts, principles and theories in: chemistry of biological molecules and their processes
- C16 Demonstrate knowledge and understanding of essential facts, concepts, principles and theories: principles and procedures in chemical engineering
- C17 Demonstrate knowledge and understanding of essential facts, concepts, principles and theories in: metrology of chemical processes including quality management
- C18 Demonstrate knowledge and understanding of essential facts, concepts, principles and theories: principles of electrochemistry
- C19 Apply knowledge and understanding to solve basic problems of quantitative and qualitative nature
- C20 Evaluate, interpret and synthesize data and chemical information
- C21 Recognize and implement good scientific practices for measurement and experimentation
- C22 Process and perform computational calculations with chemical information and chemical data
- C23 Present oral and written scientific material and scientific arguments to a specialized audience
- C24 Recognize and analyze new problems and plan strategies to solve them
- C25 Handle chemicals safely, considering their physical and chemical properties, including the evaluation of any specific risks associated with its use
- C26 Perform common laboratory procedures and use instrumentation in synthetic and analytical work
- C27 Monitor, by observation and measurement of physical and chemical properties, events or changes, and document and record them in a consistent and reliable way
- C28 Interpret data derived from laboratory observations and measurements in terms of their significance and relate them to the appropriate theory
- C29 Demonstrate skills for numerical calculations and interpretation of experimental data, with special emphasis on precision and accuracy
- D1 Communicate orally and in writing in at least one of the official languages of the University
- D2 Communicate at a basic level in English in the field of chemistry
- D3 Learn independently
- D4 Search and manage information from different sources
- D5 Use information and communication technologies and manage basic computer tools
- D6 Use mathematics, including error analysis, estimates of orders of magnitude, correct use of units and data representations
- D7 Apply theoretical knowledge in practice
- D8 Teamwork
- D9 Work independently
- D10 Work at a national and international context
- D11 Adapt to new situations
- D12 Plan and manage time properly
- D13 Make decisions
- D14 Analyze and synthesize information and draw conclusions
- D15 Evaluate critically and constructively the environment and oneself
- D16 Develop an ethical commitment
- D17 Develop concern for environmental aspects and quality management
- D18 Generate new ideas and show initiative

Learning outcomes

Expected results from this subject

Training and Learning Results

C1 C2	D1 D2
C2 C3	D2 D3
C4	D4
C5	D5
C6	D6
C7	D7
C8	D8
C9	D9
C10	D10
C11 C12	D11 D12
C12 C13	D12 D13
C14	D13
C15	D15
C16	D16
C17	D17
C18	D18
C19	
C20	
C21	
C22 C23	
C23	
C25	
C26	
C27	
C28	
C29	

Contents

Topic

(*)Dado o seu carácter especial, a materia non ten contidos propios.

	Class hours	Hours outside the classroom	Total hours
Projects	160	256	416
Jobs and projects	0.5	33.5	34

Methodologies	
	Description
Projects	Individual work done by the students under the supervision of one or two tutors. The assignment of the subject will be done following the TFG norms approved by the Faculty of Chemistry.

Description

Personalized attention

Methodologies

Projects

Assessment

Description

Qualification Training and Learning Results

Projects	Evaluation by the tutor of the competences achieved during the realization of the work assigned, in accordance with the criteria established and published previously.	30	A1 A2 A3 A4 A5	C1D1C2D2C3D3C4D4C5D5C6D6C7D7C8D8C9D9C10D10C11D11C12D12C13D13C14D14C15D15C16D16C17D17C18D18C19C20C21C22C23C24C25C26C27C28
Jobs and projects	Evaluation by a jury in public session, in accordance with criteria established and published previously.	70	A1 A2 A3 A4 A5	$\begin{array}{cccc} C29 \\ C1 & D1 \\ C2 & D2 \\ C3 & D3 \\ C4 & D4 \\ C5 & D5 \\ C6 & D6 \\ C7 & D7 \\ C8 & D8 \\ C9 & D9 \\ C10 & D10 \\ C11 & D11 \\ C12 & D12 \\ C13 & D13 \\ C14 & D14 \\ C15 & D15 \\ C16 & D16 \\ C17 & D17 \\ C18 & D18 \\ C19 \\ C20 \\ C21 \\ C22 \\ C23 \\ C24 \\ C25 \\ C26 \\ C27 \\ C28 \\ C29 \\ \end{array}$

Other comments on the Evaluation

TFG is ruled by the norms approved in the Junta de Facultad and published in the web page web of the faculty. The TFG Commission will do public, with sufficient advance, the criteria of evaluation that will use the tutor and the jury. The TFG Commission will do public, with sufficient advance, the conditions for the written report and the public defences. All the information generated by the TFG Commission will be included in the platform Tem@ and/or in the web page of the faculty.

Sources of information

Recommendations

Subjects that are recommended to be taken simultaneously Environmental chemistry/V11G200V01902

Environmental chemistry/V11G200V01902 Pharmaceutical chemistry/V11G200V01903 Industrial chemistry/V11G200V01904